

## SOUTHERN IONICS INCORPORATED (SII) SAFETY DATA SHEET

SDS NO. 216 Effective Date: April 30, 2015 Revision Date: September 25, 2020

I. Product and Company Information						
SII Product Name(s):	AQUA-CAT® Aqua Ammonia (5 % - 19 %)	Synonym: Ammonia Solution, Aqu Ammonia				
Chemical Name:	Ammonium Hydroxide	CAS Number:	1336-21-6			
Manufacturer's Name:		<b>Emergency Contacts:</b>				
Southern Ionics Incorpo	rated	Afterhours (Sout	Afterhours (Southern Ionics):			
579 Commerce Street		1-888-610-2379				
West Point, MS 39773		For Chemical Emergency, Spill, or Accident				
Customer Service: 1-800	0-953-3585	Call CHEMTREC at 1-800-424-9300				
Web Site: www.southe	rnionics.com	CHEMTREC CCN	- 20596			

II. Hazard Identification					
OSHA HCS / GHS Classification(s):			Hazard Statement(s):		
Acute Toxicity, Oral, Cat	egory 4		Harmful if swallowed.		
Skin Corrosion, Categor	y 1		Causes severe skin burn.		
Serious Eye Damage, Ca	tegory 1		Causes serious eye damage.		
Specific Target Organ To	oxicity, Respirat	tory - single	May cause respiratory irritation.		
exposure, Category 3					
Simple Asphyxiants			May displace oxygen and cause rapid suffocation.		
Acute Aquatic Toxicity,	Category 1		Very toxic to aquatic life.		
Signal Word:	Precautionar	y Statement(s):			
Danger	Prevention:	Wash affected body	parts thoroughly after handling.		
Symbols:		Do not eat, drink, or	r smoke when using this product.		
		Wear eye and face	protection.		
le de la constant de		Wear protective gloves and clothing.			
不 影		Do not breathe mis	t, vapors, or spray.		
		Avoid release to the	e environment.		
× ×	Response:	IF SWALLOWED: R	inse mouth. Do not induce vomiting. Immediately		
NV.		seek medical advice	2.		
⟨ ₹2 ⟩		IF ON SKIN: Immed	liately remove all contaminated clothing. Rinse skin		
		with water.			
•		IF IN EYES: Rinse cautiously with water for several minutes. Remove			
		contact lenses, if present and easy to do so. Continue rinsing.			
		IF INHALED: Remove victim to fresh air and keep comfortable for			
		breathing.			
		Reference Section VI. Accidental Release Measures to collect spillage.			
		For specific treatme	ent, see Section IV. First Aid Measures.		

III. Composition / Information on Ingredients					
Chemical Name(s): CAS Number(s): %					
Ammonia (NH <sub>3</sub> )	7664-41-7	5 - 19.9			
Water 7732-18-5 Balance					

IV. First Aid Measures				
Eyes:	Immediately flush eyes with running water for at least 15 minutes, keeping eyelids open. Remove any contact lenses. Seek medical attention, if you feel unwell.			
Dermal / Skin:	Remove contaminated clothing and wash exposed area thoroughly with soap and water. Seek medical attention, if you feel unwell.			
Inhalation:	Move to fresh air immediately. If breathing is difficult, give oxygen. Seek medical attention, if you feel unwell.			
Ingestion:	If swallowed, DO NOT induce vomiting. Rinse mouth. Seek medical attention, if you feel unwell.			

V. Fire Fighting Measures						
NFPA Hazard Rating:	Health (Blue) Fire (Red) Reactivity (Yellow) Special Instructions (Whi					
	3	1	0	None		
NFPA Hazard Clas	sification: 0 = I	Least 1 = Slig	ght 2 = Moderate 3	s = High 4 = Extreme		
Extinguishing Media:	Use extinguishir	ıg media appro	opriate for surrounding	g fire (Not CO <sub>2</sub> ).		
Special Firefighting	Wear full prote	ctive clothing	and a self-contained	d breathing apparatus (SCBA)		
Procedure:	because toxic fumes are emitted. Stop flow if possible. Use water to keep fire- exposed containers cool and to protect persons shutting off flow of liquid. For a serious leak, use fire hose with a fog nozzle and plenty of water to absorb ammonia vapors.					
Unusual Fire and Explosive Hazards:	At elevated temperatures, aqua ammonia will emit ammonia gas and possibly small amounts of nitrogen oxides, which have been classified as toxic. Presence of oil or other combustible materials increases the fire hazard of ammonia gas. Ammonia concentrations in the range of 16-25 % by volume in air can be ignited or caused to explode if heated to the auto-ignition temperature.					

VI. Accidental Release Measures					
Precaution if Spilled or Released:	Steps should be taken to contain spilled liquids and prevent discharges to streams or sewer systems. Ventilate spill or leak area to disperse gas. Eliminate all sources of ignition. Stop flow if possible. If small spill, either allow it to vaporize or absorb the vapor in water. If large spill, spray the vapor cloud with water to reduce fire and fume hazard.				
<b>Neutralizing Chemicals:</b>	Neutralization with acid not recommended. Flush area with water.				

VII. Handling and Storage						
Handling:	Handle all chemicals with respect. Keep separated from incompatible substances. Handle only with equipment, materials, and supplies specified by their					
	manufacturer as being compatible and appropriate for use with this product.					
Storage:	Storage in specially designated areas outside or in detached structure is preferred.					
	Store inside only in a cool, well-ventilated area free from combustibles and away from all sources of ignition. Protect containers from corrosion and mechanical					
	damage. Containers should have safety relief valves. Keep separate from other					
	chemicals, particularly oxidizing gases, organic materials, chlorine, bromine, iodine,					
	mercury, and acids. Post visible warning signs in the storage area listing emergency					
	measures. Water hoses should be readily available to knock down vapors from spill.					

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VIII. Exposure Control / Personal Protective Equipment					
Component Workplace Control Parameters:					
Components:	CAS Number	Value	<b>Parameters</b>	Basis	
Ammonia NH <sub>3</sub>	7664-41-7	TWA	25 ppm	as Ammonia NH3 (ACGIH)	
<b>Engineering Controls:</b>	Provide exhaus	t ventilat	ion or other e	ngineering controls to ke	ep the airborne
	concentrations of	of vapors l	below their resp	pective threshold limit valu	e.
General Hygiene:				ing this material, especiall	y before eating,
	drinking, smokii	ng, or usir	ng the toilet.		
Personal Protection Equ	uipment:				
Eye:	Wear chemical g	goggles ar	nd face shield u	nless protected by a respir	ator with a full-
	_			they may trap fumes agai	nst the eyes and
	can make flushir				
Skin:	The use of gloves, boots, and aprons impermeable to the specific material handled				
	-		-	eoprene, and Viton) is adv	vised to prevent
	skin contact, pos				
Respiratory:	-			When conditions warrant	=
	NIOSH-approved	d respirat	or and cartridge	e for particulates and amm	onia.
Other Protective Items:	_	-		resistant protective clothin	_
	required. Ensure that eyewash stations and safety showers are proximal to the				
	work location.				
HMIS Classification:	Health (Blue)	Flamm	ability (Red)	Physical Hazard (Yellow)	PPE (White)
	3	-	1	0	See Above
Hazard Classifica	Hazard Classification: 0 = Minimal 1 = Slight 2 = Moderate 3 = Serious 4 = Severe				

IX. Physical and Chemical Properties					
Physical State:	Liquid	рН:	> 12		
Appearance:	Clear, colorless liquid	Molecular Weight:	35.05		
Odor:	Pungent odor	Odor Threshold:	1 - 50 ppm		
Specific Gravity: (H <sub>2</sub> O=1)	0.98 (5 % Solution) 0.94 (15 % Solution) 0.93 (19 % Solution) @ 60 °F / 15.5 °C	Weight per Gallon:	8.17 (5 % Solution) 7.87 (15 % Solution) 7.76 (19 % Solution) lbs @ 60 °F / 15.5 °C		
Vapor Density: (Air=1)	0.60 @ 32 °F (0 °C)	Vapor Pressure:	78 mm Hg (5 %) 194 mm Hg (15 %) 264 mm Hg (19 %) @ 77 °F / 25 °C		
Boiling Point: at 14.7 psia	177 °F / 80.5 °C (5 %) 120 °F / 48.9 °C (19 %)	Freezing/Melting Point:	25 °F / -4 °C (5 %) -30 °F / - 34 °C (19 %)		
Lower Explosive Limit: Flash Point:	16 % by volume Ammonia gas Not Applicable	Upper Explosive Limit: Autoignition Temp:	25 % by volume Ammonia gas 1,204 ° F 651 °C (vapor)		
Solubility in Water:	100 %	Other:	1,207 1 001 C (vapor)		

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X. Stability and Reactivity Data					
Chemical Stability: Product is stable under normal or expected use.					
Conditions to Avoid:	Heat, sunlight, incompatibles, sources of ignition.				
Incompatible Materials:	Corrosive to copper, brass, silver, zinc, aluminum alloys, and galvanized steel. Immediately boils when mixed with acids and is dangerous. Forms explosive compounds with calcium hypochlorite, bleaches, gold, mercury, silver, chlorine, and other halogens.				
Hazardous Products of Decomposition:	Burning may produce ammonia and nitrogen oxides.				

XI. Toxicological Information						
Routes of Entry:	🛛 Eyes 🔀 Skin	\overline{\text{Syes}}   \overline{\text{Skin}}   \overline{\text{Skin}}   \overline{\text{Ingestion}}   \overline{\text{Inhalation}}   align*				
Sign and Symptoms of		Burning of the eyes, conjunctivitis, skin irritations, swelling of the eyelids and lips,				
Exposure:	1	0 .	the throat, and cough	S		
	_	•	igns and symptoms of	0 0		
	ultimately, death fr	om respiratory failur	e due to pulmonary edo	ema may occur.		
Eye Contact:	Vapor is irritating t	o the eyes. Liquid wil	l cause burns.			
Ingestion:		<b>U</b> 1	h, throat, stomach, th	-		
		0	wed by vomiting of bloom	<b>,</b> ,		
	loose stools contair	ning blood. Ingestion o	of 3 - 4 mL may be fatal			
Skin Contact:	Absorption: Becaus	se of its alkalinity and	d water solubility, amn	nonia tends to break		
	down and disrupt the outer cell layers, permitting rapid penetration; however,					
	ammonia is not a systemic poison, and the effects will be limited to local effects.					
	Contact: Causes smarting of the skin and first-degree burns on short exposure. May					
	cause second-degree burns on long exposure.					
Inhalation:	Ammonia vapors are highly irritating to throat at approximately 400 ppm. Causes					
	edema, dyspnea, bronchospasm, chest pain, pink frothy sputum. Inhalation of $\geq$ 500					
	ppm ammonia is considered immediately dangerous to life and health (OSHA).					
Carcinogenicity: Not Liste	ted NPT Not Listed IARC Not Listed OSHA Not Regulated					
Ingredient Name:	Species	Test	Period	Results		
Ammonium Hydroxide	Rat 350 mg/kg Oral LD50			LD50		
Comments:						

XII. Ecological Information								
Ingredient Name:	Species Test Period Results							
Ammonia NH <sub>3</sub>	Chinook Salmon	Chinook Salmon 0.45 mg/L 96 hrs LC50						
Comments:	via combination with to soil, sediment p Biodegradation of a	relatively quickly in an an sulfate ions or washoparticles, and colloid mmonia to nitrate of the ological oxygen deman	out by rainfall. Ammo ds in water under ccurs in water undei	nia strongly adsorbs aerobic conditions.				

XIII. Disposal Considerations		
Waste Disposal:	Always dispose of material in accordance with local, state, and federal regulations.	

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XIV. Transportation Information							
Proper Shipping Name:	Ammonium Hydroxide, with more than 10 % but not more than 35 % as ammonia.						
	Marine polli	utant.					
DOT Classification:	8						
<b>Identification Number:</b>	UN 2672	Packing Group:	III	Other Labels:	Corrosive		
Comments:							

XV. Regulatory Information						
Inventory Status:		US Regulations:				
U. S. TSCA	Yes	SARA 302 TPQ 500 lbs as ammo		onia NH <sub>3</sub>		
Europe EINECS	Yes	SARA 304 RQ 100 lbs as ammonia NH <sub>3</sub>				
Canadian DSL	Yes	SARA 313 List Listed				
Japan ENCS	Yes	CERCLA (RQ) 1,000 lbs for pure ammonium hydroxide			<del>S</del>	
Korean KECI	Yes	RCRA 261.33 Not Listed				
Philippines PICCS	Yes					
Australian AICS	Yes	SARA 311/312 Acute Chronic Fire Release of Pressure Reactive			ressure 🗌 Reactive	
International Regulations:				Other Regulations:		
Canada WHMIS	Е	Corrosive		California PROP 65	Not listed	
EINECS	231-635-3	as Anhydrous Amn	nonia			
EINECS	215-647-6	as Aqua Ammonia				

XVI. Other Information			
Other:			
Revision Notes:	05.20.16 SDS product compositoin was changed from 19 - 30.5 to 5 - 19. 06.02.16 Added Marine pollutant designation under proper shipping name. 09.25.20 Reviewed as part of a 3-year review process. Updated logo and formatting.		
MSDS Replacements:	SII MSDS 097 AQUA-CAT® Aqua Ammonia		

## **SALES OFFICE**

## **For Product Information:**

TEL: 662-494-3055 FAX: 662-494-2828 Post Office Drawer 1217 West Point, MS 39773 **To Place an Order:** TEL: 800-953-3585 FAX: 800-953-3588

## **IMPORTANT**

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